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A New Approach to the Synthesis of the 25-Hydroxy-22-Oxa-Vitamin D3 Side Chain

Yagamare Fall

Departamento de Química Orgánica, y Unidad Asociada al CSIC, Facultad de Química, Universidad de Santiago de Compostela,

15706 Santiago de Compostela, Spain FAX 34-81-595012, E-Mail qoyago@usc.es

Abstract: An efficient new method is described for the construction of the 25-hydroxy-22-oxavitamin D3 side-chain, which is present in several analogues of calcitriol with antitumour activity. © 1997 Elsevier Science Ltd.

In recent years it has come to light that 1α ,25-dihydroxyvitamin D₃ (1, calcitriol), the hormonally active form of vitamin D₃ in calcium homoeostasis,¹ can induce cell differentiation and inhibit cell proliferation, which suggests its possible use in the treatment of certain cancers and skin disorders.² Since 1α ,25-dihydroxyvitamin D₃ itself is of limited therapeutic value due to its hypercalcemic effects,² interest in this field has focused on the development of analogues having strong cell differentiating and weak calcemic effects.³ Hitherto, several interesting analogues with a modified side chain have been synthesized (Figure 1), among them 1α ,25dihydroxy-22-oxavitamin D₃ (**2b**),⁴ which inhibits the growth of certain cancers⁵ and is a potent immunomodulator *in vivo* but does not cause hypercalcaemia.⁶ In this paper we report a rapid and practical method for the construction of the 25-hydroxy-22-oxavitamin D₃ side-chain which is present not only in **2b** but also in several other calcitriol analogues with antitumour activity.

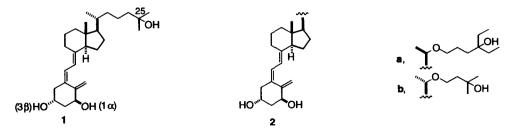
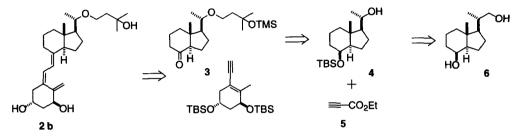


Figure 1. Structure of 10,25-dihydroxyvitamin D3 and some of its analogues modified at the side chain.

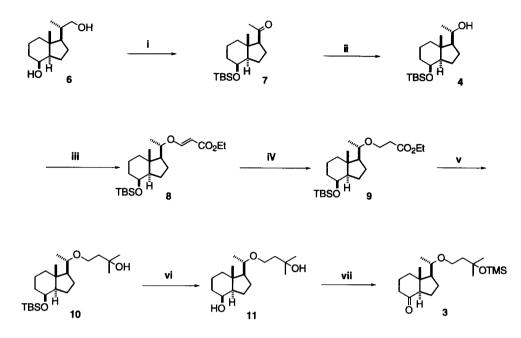
As outlined in Scheme 1, our synthetic approach to **2b** takes advantage of the readily available Inhoffen-Lythgoe diol (6) and makes use of the well-established convergent dienyne route to vitamin D analogues.⁷ The success of the proposed synthesis hinged on the viability of the reaction between ethyl propiolate and the alcohol **4**, which was envisaged to afford **3**, the precursor of **2b**.



Scheme 1. Retrosynthetic plan.

Diol **6**, which was obtained by degradation of vitamin D₂,⁸ was converted to ketone 7^9 in 55% yield by sequential monotosylation, protection, conversion to the aldehyde by the modified Kornblum procedure,⁹ and oxidation (Scheme 2). Ketone **7** was then stereoselectively reduced with an alkali metal to obtain alcohol **4**.¹⁰ To our great satisfaction, room temperature reaction of alcohol **4** with ethyl propiolate in the presence of *N*-methylmorpholine in benzene¹¹ cleanly afforded the desired alkoxyacrylate **8**¹² in 85% yield. Conversion of **8** to the ester **9** (80%) by hydrogenation on Pd/C followed by treatment with methylmagnesium chloride (MeMgCl) in Et₂O gave an 80% yield of the tertiary alcohol **10**,¹² which bears the desired 25-hydroxy-22-oxavitamin D3 side chain. The silyl protecting group of **10** was removed by reaction with HF in acetonitrile at room temperature to give the diol **11**¹² (91%). Oxidation of **11** with pyridinium chlorochromate, followed by overnight treatment with 1-(trimethylsilyl)imidazole in THF at room temperature, afforded the target compound **3**¹² (94% for the two steps, 15% from **6**), which can be converted to 1α ,25-dihydroxy-22-oxavitamin D₃ (**2b**) by known synthetic routes.⁸

This new method for construction of the 25-hydroxy-22-oxavitamin D₃ side chain compares favourably with previous methods as regards to efficiency and rapidity. Furthermore, the ester **9** is a useful intermediate in the synthesis of other C-25 vitamin D analogues, the synthesis of which is in progress.



Scheme 2. Reaction conditions. (i) *p*-TsCl, py (95%); *t*-BuMe₂SiCl, imidazole (83%); NaHCO₃ (1 equiv) DMSO, rt (85%); O₂, *t*-BuOK, *t*-BuOH, rt (82%). (ii) K, *i*-PrOH reflux (60%) (iii) Ethyl propiolate, *N*-methylmorpholine, PhH, rt, 12 h (85%). (iv) H₂, Pd/C, EtOH, (80%). (v) MeMgCl, Et₂O, -10 °C to rt (80%). (vi) HF, CH₃CN (91%). (vii) PCC, CH₂Cl₂, rt; TMS-imidazole, THF, rt, 12 h (94%).

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- 10. (a) Kirk, D. N.; Mudd. A.; J. Chem. Soc.(C) 1969, 968-974. (b) Alcohol 4 was easily separable from its C20 (R) epimer by column chromatography (Hexane/EtOAc 4%). Ratio 4(S):4(R) = 3/2. Proton NMR of the two epimers showed as expected a difference in the value of the chemical shifts of the C18 methyl groups : δ C18 = 0.916 ppm for 4(S) whereas δ C18 = 0.994 ppm for 4(R).
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- 12. All new compounds were characterized by ¹H and ¹³C NMR, combustion analysis and/or high resolution mass spectral data.

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